

$$U_{\text{rot}} = \frac{1}{2} F_{\text{rot}} T S_{\text{rot}} = NkT \quad ! \quad 2 \text{ degrees of freedom}$$

$$\text{Total energy } U = U_{\text{trans}} + U_{\text{rot}} = \frac{5}{2} NkT$$

in general when molecule are not symmetric,  $I_1, I_2, I_3$

$$\rightarrow U_{\text{rot}} = \frac{3}{2} NkT$$

$$U = U_{\text{rot}} + U_{\text{trans}} = 3NkT$$

Grand canonical ensemble.

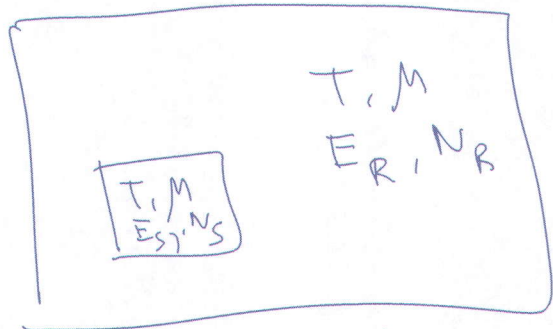
describes open system: exchange energy & particles with surrounding.

microcanonical  $(E, V, N)$  T is deduced

canonical  $(T, V, N)$  E is deduced

grand canonical  $(T, V, \mu)$  E, N is deduced

consider



where  $\Omega(E, N)$

a micro state of the system with energy  $E_S, N_S$ .

$$P = \frac{\Omega_R(E_R, N_R)}{\Omega(E, N)} \propto \Omega_R(E - E_S, N - N_S)$$

$$\text{expand } k \ln \Omega_R = k \ln \Omega_R(E, N) - \frac{\partial}{\partial E} (k \ln \Omega) \cdot E_S - \frac{\partial}{\partial N} (k \ln \Omega) \cdot N_S$$