

Problem II.1

The change in entropy, when particles flow is given by

$$\Delta S = \left(\frac{\partial S_1}{\partial N} - \frac{\partial S_2}{\partial N} \right) \Delta N, \geq 0$$

But since

$$\frac{\partial S}{\partial N} = -\frac{\mu}{T}$$

then
$$\Delta S = \frac{1}{T} (\mu_2 - \mu_1) \Delta N, \geq 0$$

considering temperature to be the same in both systems.

We conclude that when $\mu_2 > \mu_1$, then ΔN_1 will be positive $\Rightarrow N_1$ will increase.

\therefore particles will flow from high to low μ .

Problem III.2

Average speed:

$$\begin{aligned}\langle v \rangle &\equiv \int_0^{\infty} v f(v) dv = A \int_0^{\infty} v^2 e^{-\frac{v}{v_0}} dv = \\ &= A \left[e^{-\frac{v}{v_0}} (-2v_0^3 - 2v_0^2 v - v_0 v^2) \right]_0^{\infty} = \\ &= 2A v_0^3\end{aligned}$$

Now: $\int_0^{\infty} f(v) dv = 1 \Rightarrow [A e^{-\frac{v}{v_0}} (-v_0 - v_0^2)]_0^{\infty} = 1$

↑
normalization
condition

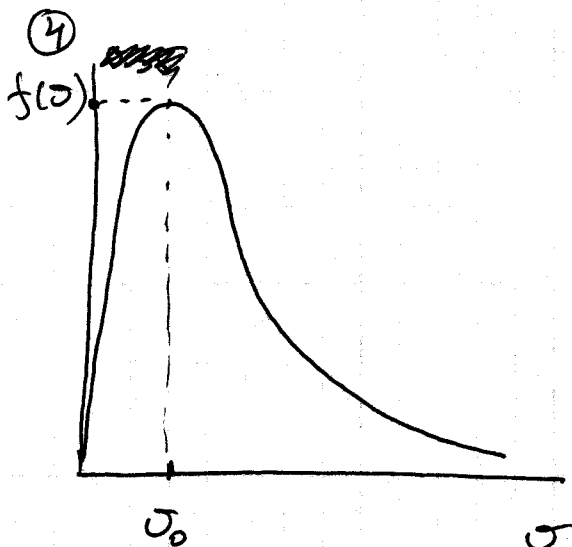
① $A = \frac{1}{v_0^2}$

② $\langle v \rangle = 2v_0$

③ Most probable speed

corresponds to the biggest probability, i.e. it is obtained from $\frac{d}{dv} f(v) = 0$, which, in principle, is different from $\langle v \rangle$. In our case

$$v_{mp} = v_0$$



Problem II.3

NOTE: Several solutions are possible.

The time intervals of such a system being in the homogeneous and separated states (t_0 and t_1) have the same ratio as corresponding probabilities.

$$\frac{t_0}{t_1} = \frac{w_0}{w_1} = \exp(S_0 - S_1), \text{ since } w \propto e^S$$

On the other hand: $S_0 - S_1 = \ln 2^{2N}$

$$\Downarrow$$
$$\frac{t_0}{t_1} = 2^{2N}$$

The time t_1 is the time for a separated state to be destroyed. We can estimate it using $t_1 \sim \frac{l}{\langle v \rangle}$, where $l \sim 10 \text{ cm}$ is a typical size of the vessel, and $\langle v \rangle \sim 10^3 \frac{\text{m}}{\text{s}}$ typical molecular velocity at room temperature. $t_0 \sim 10^{10} \text{ years} \sim 10^{17} \text{ s}$.

$$\therefore N \approx \frac{\log_2 \frac{t_0}{t_1}}{2 \log_2 2} \approx 35.$$

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This is one of the possible solutions!